

Idaho National Engineering and Environmental Laboratory

Fundamental UNEX Chemistry

Project #81995

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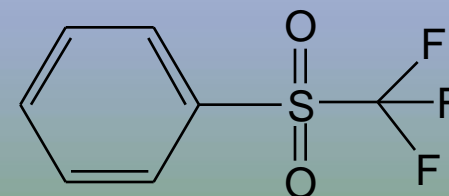
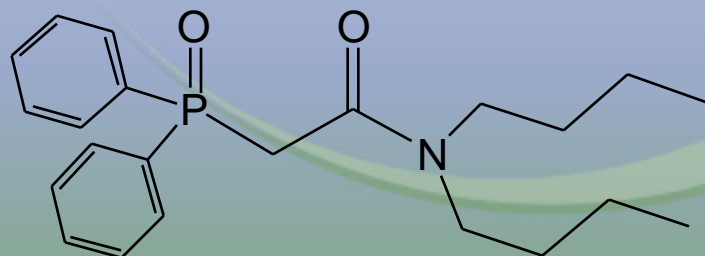
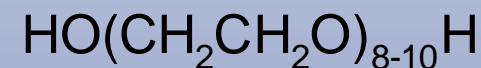
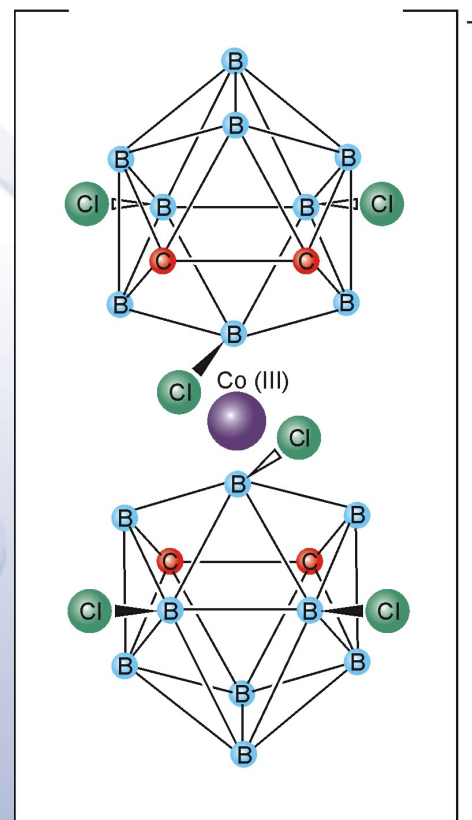
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Background: The UNEX Solvent System

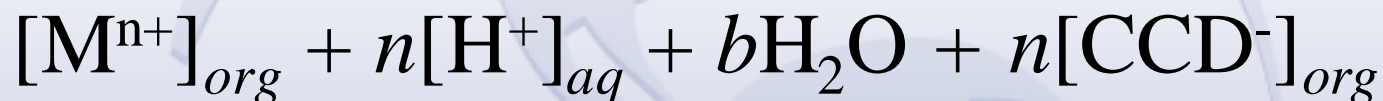
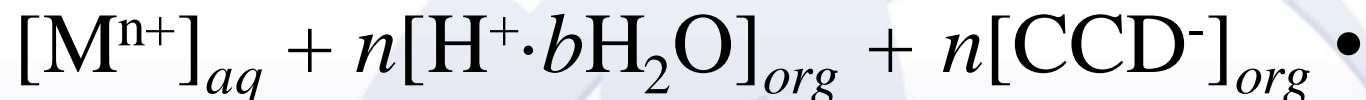
- Chlorinated Cobalt Dicarborollide (CCD) for Cs extraction.
- Polyethylene glycol (PEG-400) for Sr extraction.
- Diphenyl-N,N'-di-n-butylcarbamoylmethylphosphine oxide (CMPO) for actinide extraction.
- Current diluent is phenyltrifluoromethyl sulfone (FS-13).



RESEARCH OBJECTIVES

- ***UNEX Process indeed works, but how? - what fundamental chemistry occurs in the process?***
- ***The objective of this project is to define the underlying chemical phenomena (mechanisms) operative during the extraction process including-***
 - ***stoichiometry of extracted species***
 - ***effects of water, nitrate, and acid***
 - ***solvent effects***

Extraction in CCD Systems

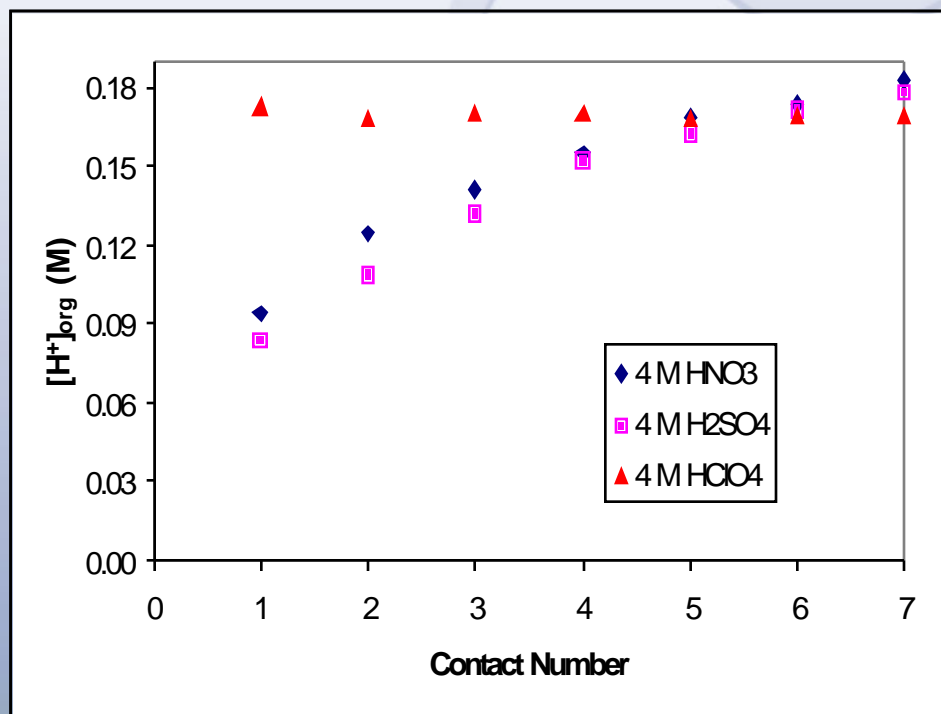


Acid (4 M)	<i>b</i>
HClO₄	5.3 ± 0.04
HNO₃	5.7 ± 0.06
H₂SO₄	5.4 ± 0.03
^a Literature	5.5

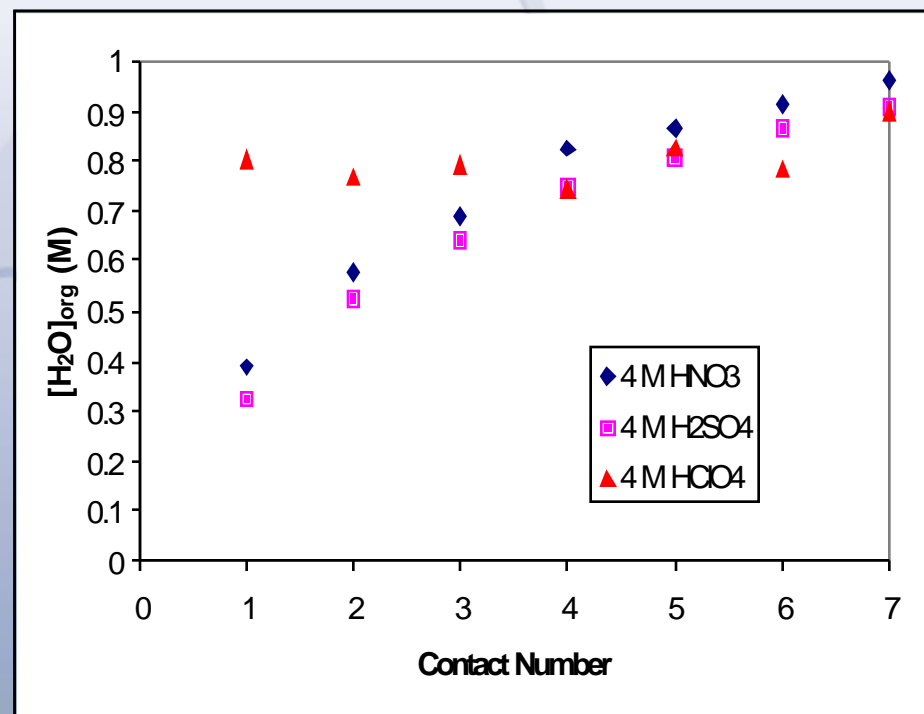
^a Vanura, P.; Makrlik, E.; Rais, I.; Kyrs, M. *Coll. Czech. Chem. Commun.* **1982**, Vol. 5 (47), 1444-1464

Cs Extraction in CCD Systems - Organic Phase Acid/Water Content

Initial Org. Phase: ~ 0.16 M CsCCD in FS-13



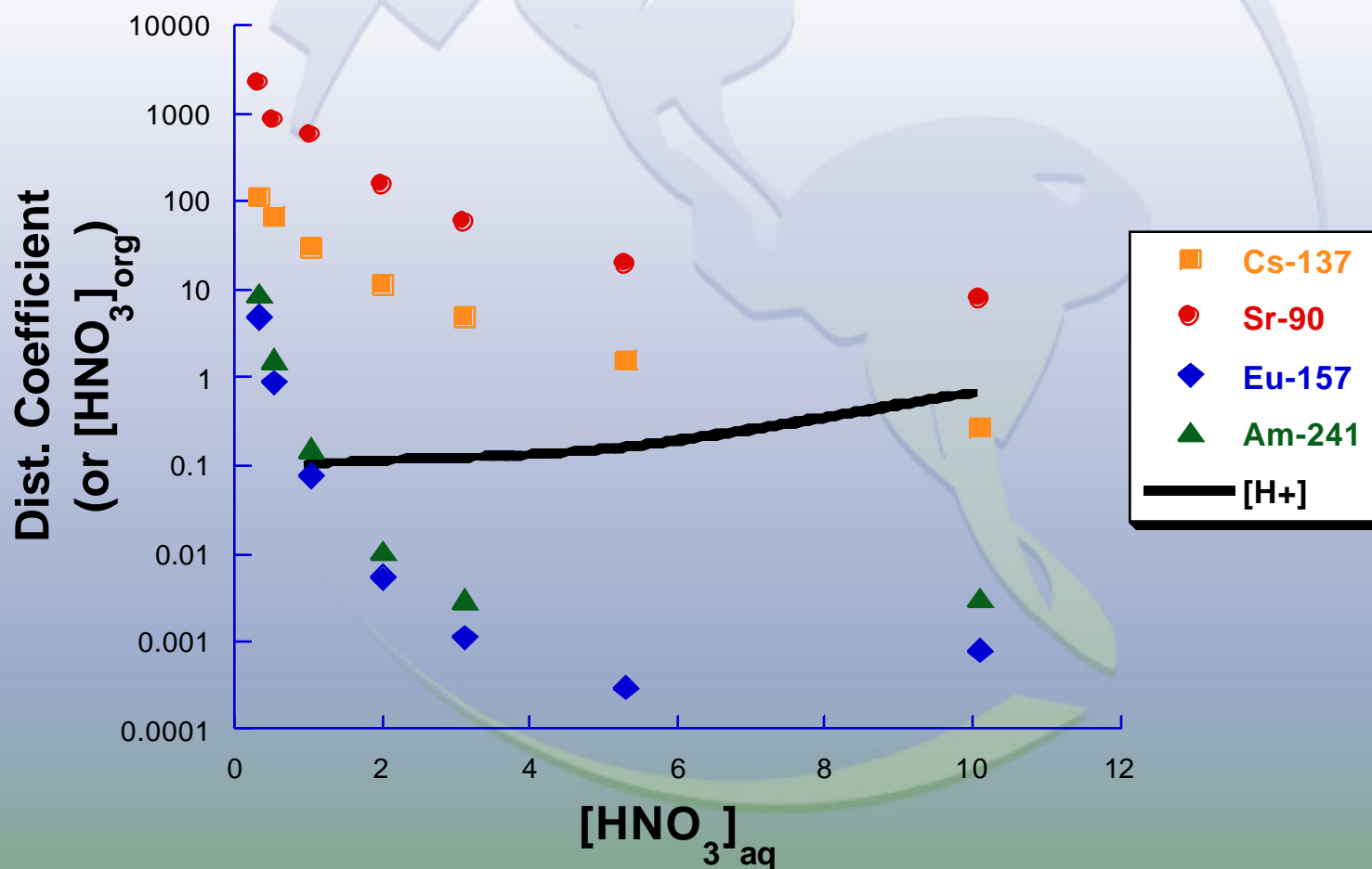
**Acid concentration
(Potentiometric Titration)**

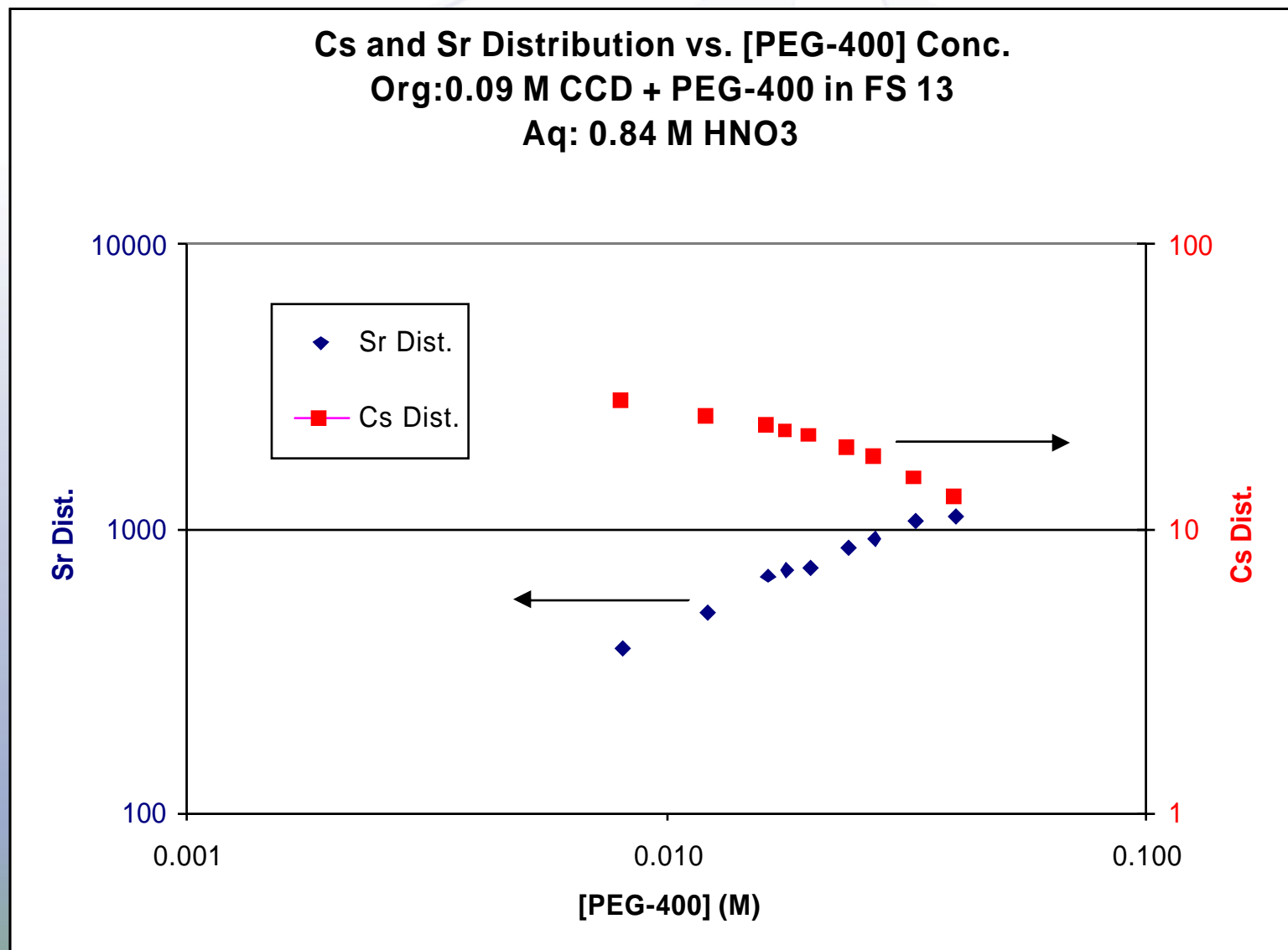


**Water concentration
(Karl Fischer Titration)**

Extraction Dependency on Aqueous $[HNO_3]$

Organic Phase: 0.08 M CCD + 0.016 M PEG-400 in FS-13 (25 °C)



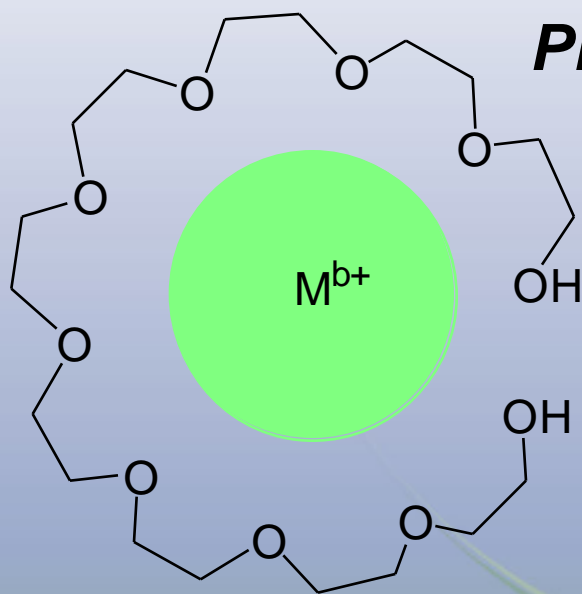


Organic Phase Strontium Complex

- ***IR spectroscopic studies of the organic phase:***
 - ***Sr⁺² saturated solution of
0.05 M CCD in DCE (w/o PEG):
[Sr⁺²•8H₂O] CCD₂⁻***
 - ***Sr⁺² saturated solution of
0.05 M CCD & 0.05 M PEG-400 in DCE:
[Sr⁺²•PEG] CCD₂⁻ 1:1 Sr:PEG Complex***

Organic Phase Strontium Complex

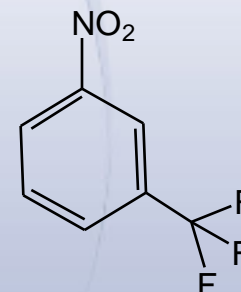
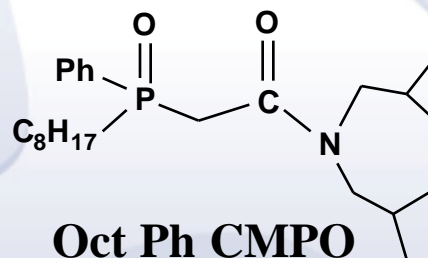
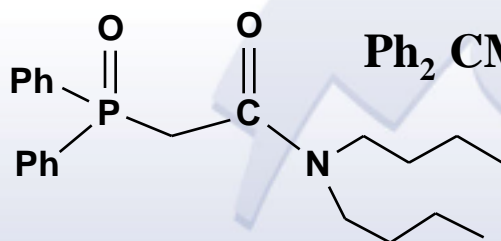
Ethereal oxygen's in PEG displace 8 H₂O molecules from 1st coordination sphere. Consistent with Sr⁺² extraction data vs. PEG size:



	Ave. n	D_{Sr}
PEG-300	6-7	4.0
PEG-400	8-10	4.1
PEG-1500	33-34	0.6



ACTINIDE EXTRACTION BY CMPO:



Distribution Data

Aq: 3M HNO₃ Org: 0.005 M CMPO in F-3

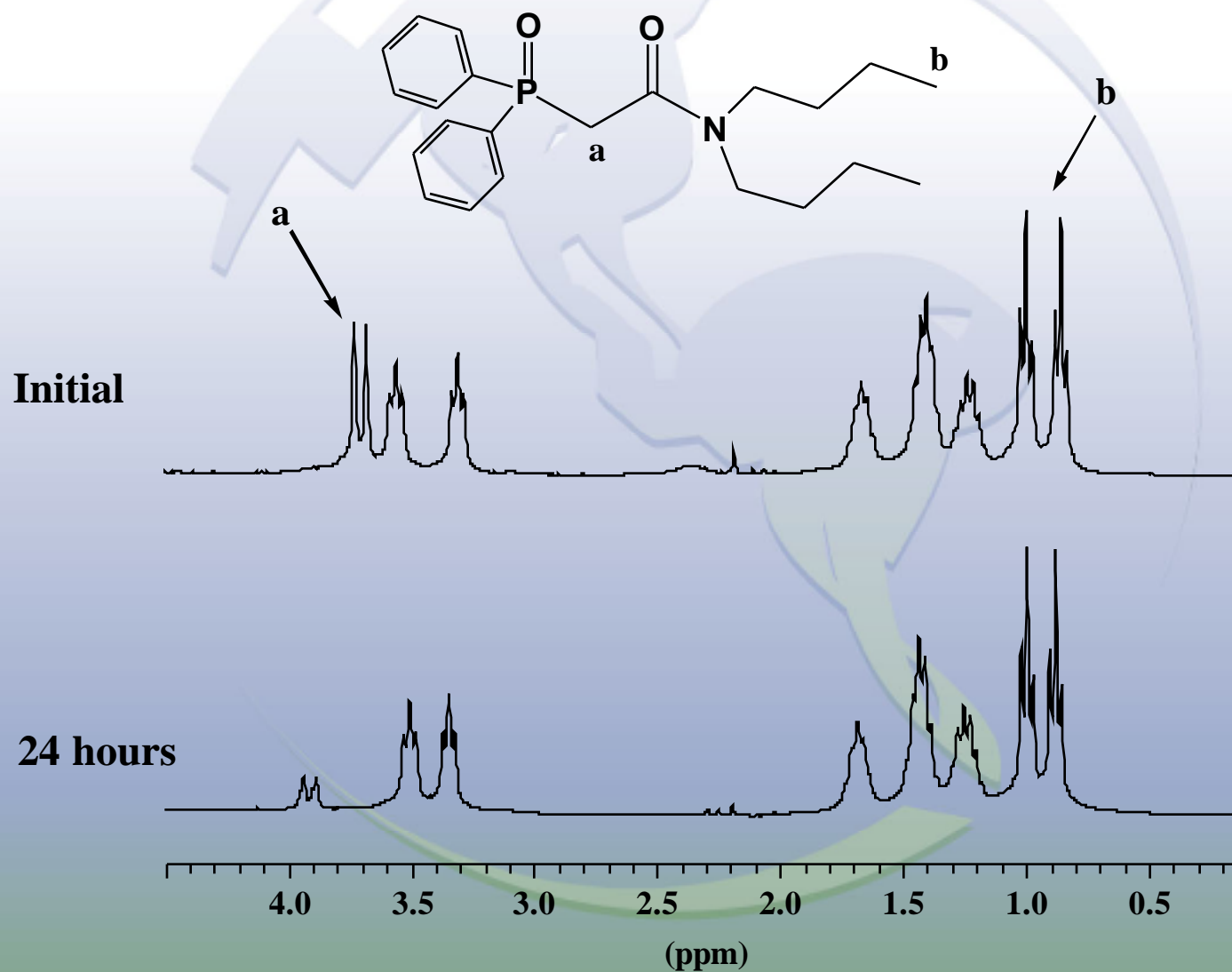
Extractant	D _{Am}	D _{Eu}	K _{HNO₃, Ex}
Oct Ph CMPO	0.05	0.03	2.0 ^a
Ph ₂ CMPO	0.45	0.30	0.76 ^a

^a Data of Horowitz, et. al. 1986

H — D Exchange

- ***Exchange of protons for deuterium in the methylene bridge observed in NMR spectra of both CMPO compounds.***
- ***Exchange initiated using 3 M HNO_3 , DNO_3 , HClO_4 , DClO_4 , $\text{LiNO}_3 + \text{La}(\text{NO}_3)_3$, or $\text{NaNO}_3 + \text{La}(\text{NO}_3)_3$, each in D_2O .***
- ***Acidity of bridge protons indicative of compound basicity.***

H — D Exchange



D_{Am} vs. H-D Exchange Rate (•)

Extractant	D_{Am}	• (hrs)
Oct Ph CMPO	0.05	24
Ph ₂ CMPO	0.45	6 - 7

SUMMARY

- In the HCCD/PEG-400 system, simultaneous Cs/Sr extraction can be represented by the following (simplified) reactions:***



- H-D exchange in the methylene bridge of CMPO has been observed. Rate of exchange qualitatively interpreted to indicate bidentate coordination of hydrated proton with CMPO.***

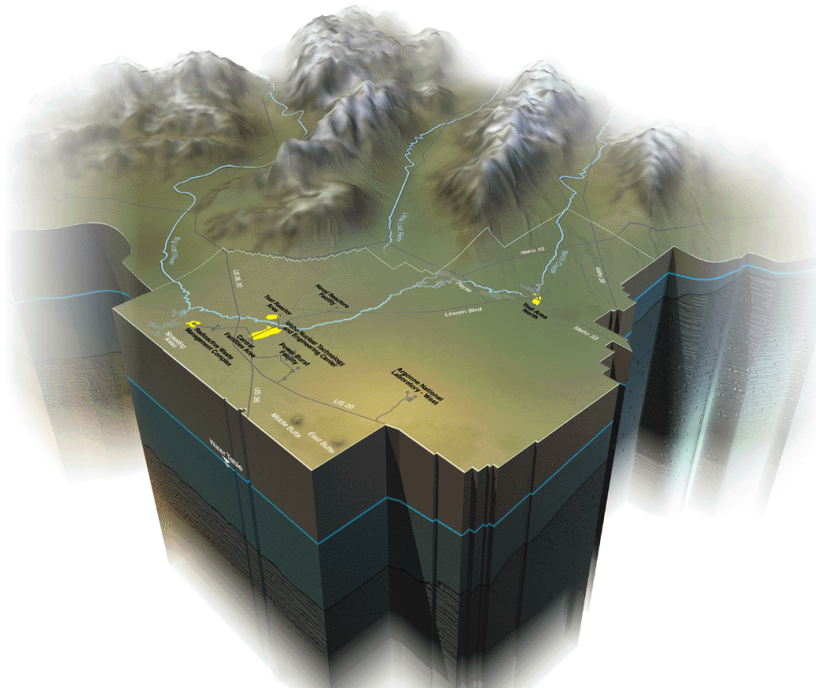
Research Highlights

- ***5 Journal Articles published/accepted/submitted thus far.***
 - ***2 drafted***
 - ***Several others planned***
- ***6 Technical presentations at National/International meetings***
- ***1 Graduate student supported***
- ***Cs/Sr Extraction results used to define solvent composition for spent nuclear fuel treatment in the DOE NE Advanced Fuel Cycle Initiative***

Future Research Direction

- ***Emphasis in FY-05 (final year of project) on further evaluation of actinide/lanthanide interactions with CMPO in the UNEX system:***
 - ***Non-synergistic (independent) of PEG?***
 - ***Anion stabilization (CCD^- vs. to NO_3^-)?***
- ***Verify of bidentate coordination of CMPO and hydrated H^+ in the organic phase.***
- ***Evaluation of organic phase structures via IR and NMR spectroscopic techniques.***

Acknowledgments



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